

# CRYSTAL STRUCTURES

## Lecture 1

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### 1 Crystal Structures



FIGURE 1: Crystals of native copper.

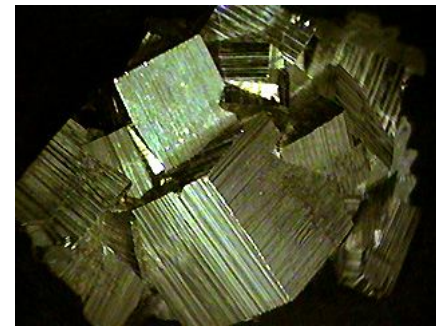


FIGURE 2: Crystals of pyrite ( $\text{FeS}_2$ ).



FIGURE 3: Crystals of quartz ( $\text{SiO}_2$ ) - the original *κρυσταλλος*.

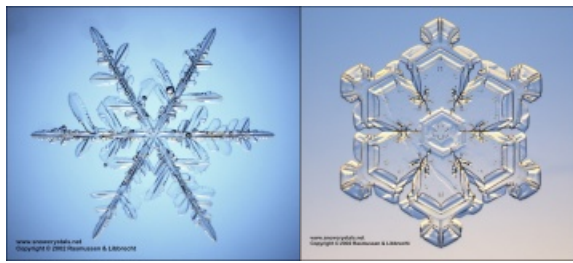


FIGURE 4: Snow crystals.

## 1.1 What is special about crystals?

- precise symmetries
- flat surfaces
- straight edges
- Haüy's "Tout est trouvé!" on dropping iceland spar

## 1.2 What does this suggest about their structure?

Regular pattern of simple building blocks (Kepler, Robert Hooke, Huygens, Descartes).



A crystal made from spherical particles, according to Robert Hooke (*Micrographia Restaurata*, London 1745).

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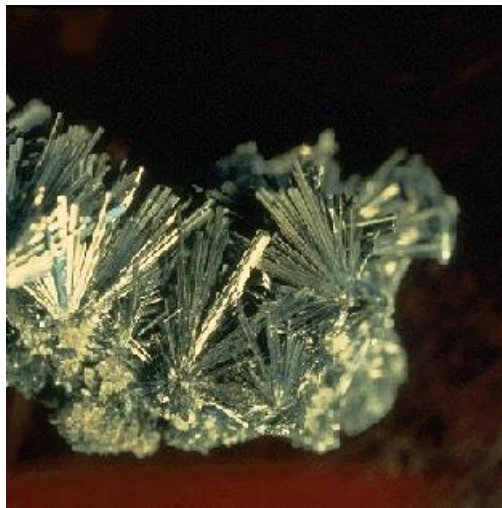
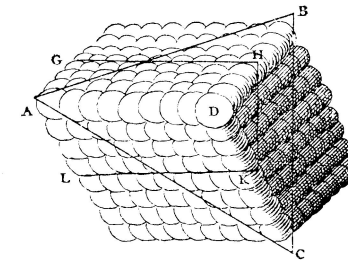


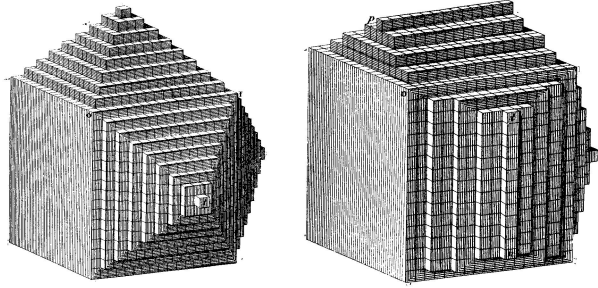
FIGURE 5: Crystals of stibnite ( $Sb_2S_3$ ).



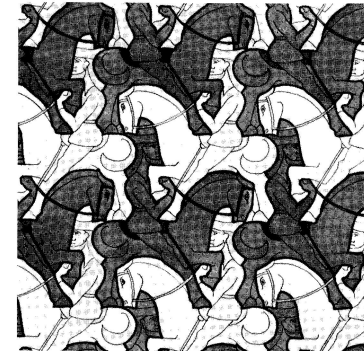
Christiaan Huygens's picture of a calcite ( $CaCO_3$ ) crystal made from spherical particles (*Traité de la Lumière*, Leiden 1690).

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A crystal structure as depicted by René Haiüy (*Traité de Cristallographie*, Paris 1822).



The plane is completely filled.

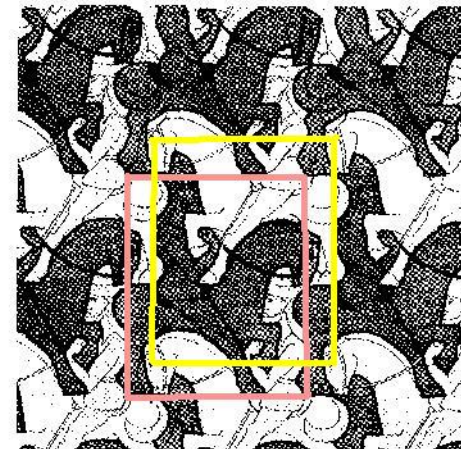
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### 1.3 Artistic Example

The key points about building this pattern are that *motifs* are assembled *periodically* the motifs are all in the same orientation note that the motif contains *two* knights

Figure shows engravings by M.C. Escher



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We can pick a unit cell: but note

- the unit cell is not unique. For example, we could pick a cell with a white knight in the middle
- or we could pick a larger cell which could be a bigger square or a rectangle or other shape

## 1.4 Formal description

Separate the motif from the repetition pattern.

### 1.4.1 The Lattice

A *lattice* is an arrangement of points in space such that the environment of any point is identical to that of any other point.

Note: points, space – this is now a *mathematical problem*.

The mathematicians tell us how many different lattice types there are in spaces of 2, 3,... dimensions. These are the *Bravais lattices*. Lattices have symmetries, more fully *point group symmetries*, described in terms of rotations and reflections.

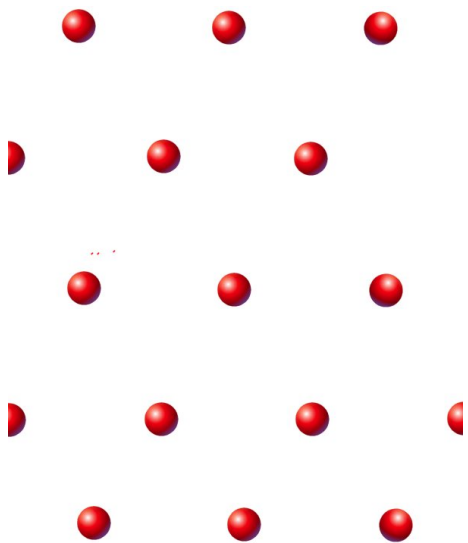
Remember: the lattice is *not* the crystal – it's the collection of points in space on which the crystal is hung (but people often use the word lattice when they mean crystal).

### 1.4.2 Lattice vectors

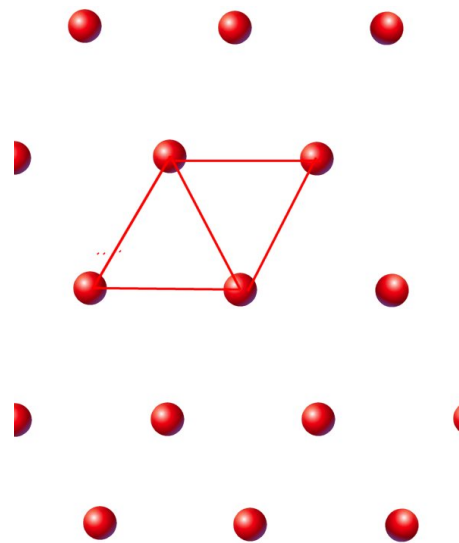
A *lattice vector* is any vector joining two lattice points. It is convenient to define a set of *primitive lattice vectors*: this is the set of the shortest linearly independent lattice vectors. Linear independence ensures that they can span all dimensions of the space - for example, in 2D they must not be parallel, and in 3D in addition they must not lie in the same plane. These vectors, conventionally referred to as  $\mathbf{a}$ ,  $\mathbf{b}$  and  $\mathbf{c}$ , allow us to start from any point on the lattice and generate the rest of the lattice points at  $n_1\mathbf{a} + n_2\mathbf{b} + n_3\mathbf{c}$  where  $n_1$ ,  $n_2$  and  $n_3$  are integers, running in principle from  $-\infty$  to  $+\infty$ .

### 1.4.3 The Unit Cell

A *unit cell* is a volume (area in 2D) which, when repeated by being translated by the lattice vectors, will fill all space. N.B. *translated*, without rotation or change of shape.



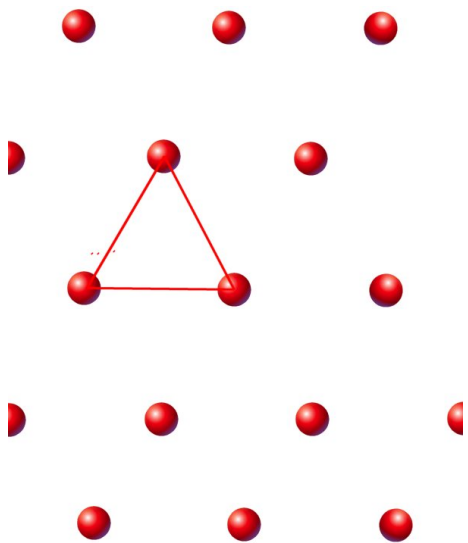
The triangular lattice - the dots represent *points*, *not atoms*.



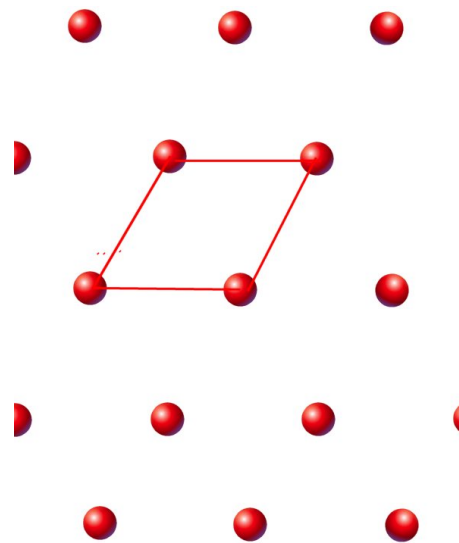
because we cannot fill space just by replicating it – we have to invert it.

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The triangle is not a suitable unit cell,



The rhombus is a suitable unit cell, the triangle is not.

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#### 1.4.4 Number of lattice points in cell

Two approaches:

Count points, sharing face, edge and corner points

Shift the cell so that all points are internal, then count

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Rectangular lattice (points have been given size to allow us to subdivide them)

Here, each point is shared with *four* neighbouring cells, so the cell contains  $4 \times \frac{1}{4} = 1$  point.

Alternatively, we can take advantage of the fact that the unit cell is not uniquely defined, so we can shift it.

This is a *primitive* unit cell.

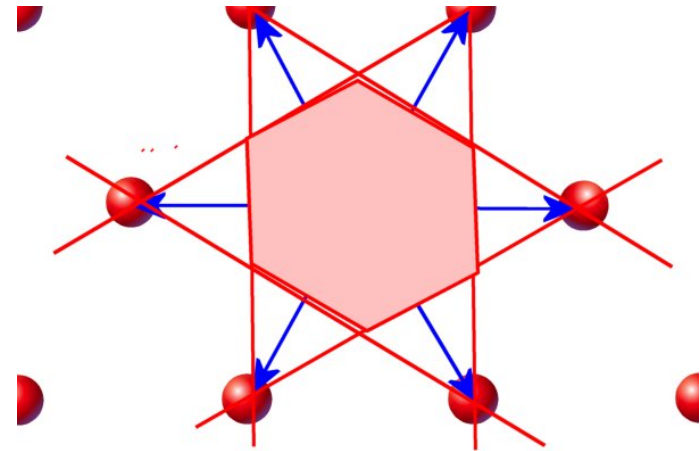
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#### 1.4.5 Wigner-Seitz cell

Construction

- select a lattice point
- draw lines joining it to its neighbours
- draw perpendicular bisectors (planes in 3D, lines in 2D) of those lines
- the Wigner-Seitz cell is the volume (area in 2D) is the area within the bisectors.

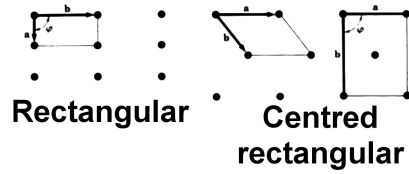
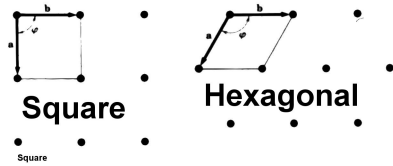
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The Wigner-Seitz cell tends to show the symmetry of the lattice.

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## 1.4.6 Five Lattices in Two Dimensions



Lattice	Unit Cell	Restrictions	Symmetry
Oblique	Parallelogram	$a \neq b, \phi \neq 90^\circ$	<b>2</b>
Square	Square	$a = b, \phi = 90^\circ$	<b>4mm</b>
Hexagonal	$60^\circ$ Rhombus	$a = b, \phi = 120^\circ$	<b>6mm</b>
Primitive Rectangular	Rectangle	$a \neq b, \phi = 90^\circ$	<b>2mm</b>
Centred Rectangular	Rectangle	$a \neq b, \phi = 90^\circ$	<b>2mm</b>

## Summary

### Definitions

Analysis of structure

Beginning on real structures